



Increasing Importance of Ion Chromatography for Pharmaceutical Analysis

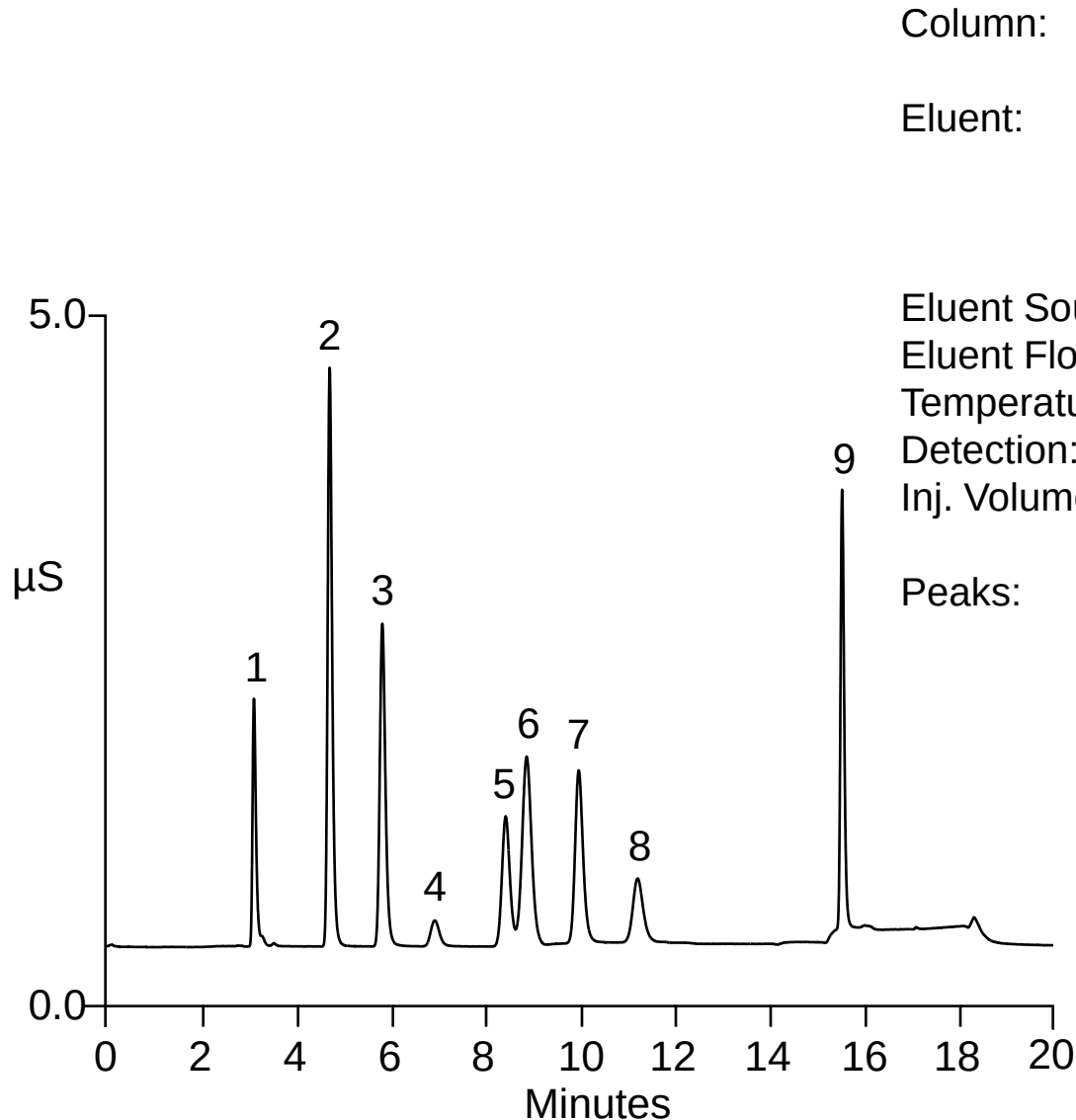
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ISC 2016, Cork, Ireland
August 31th, 2016

- Introduction to Ion Chromatography
- What Ion Chromatography Offers for Pharmaceutical Analysis
- Review of the Three IC Applications for Pharmaceutical Analysis

At the most basic level....

Ion Chromatography =
Ion-Exchange + Chemically
Suppressed Conductivity

Separation of Common Anions and TFA

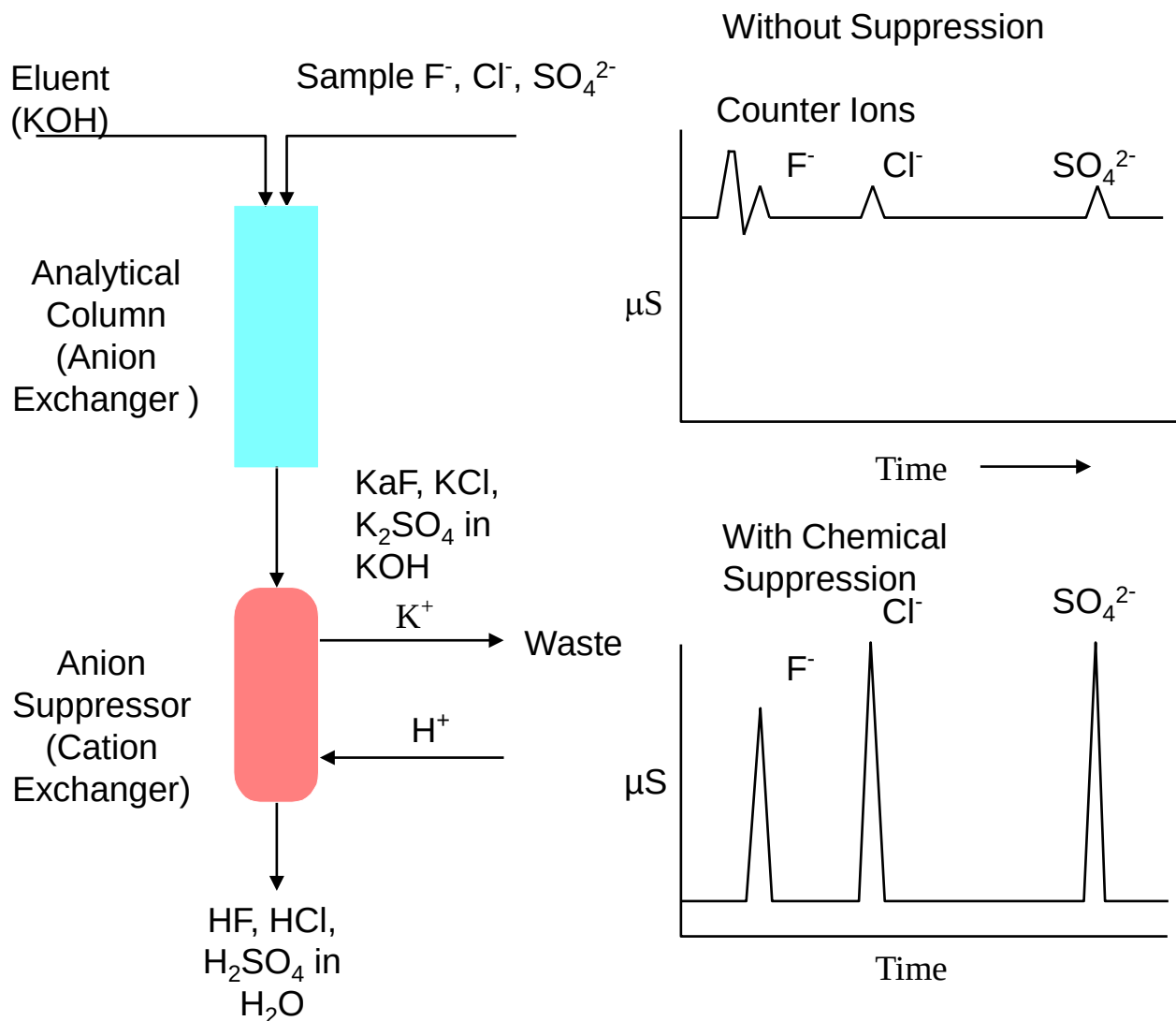


Column: Thermo Scientific™ Dionex™
IonPac™ AG18 and AS18
Eluent: 22 mM KOH for 0–6 min,
28 mM KOH from 6–12 min,
50 mM KOH for 12–15 min,
22 mM KOH from 15–20 min
Eluent Source: EG50
Eluent Flow Rate: 1.0 mL/min
Temperature: 30 ° C
Detection: Suppressed conductivity
Inj. Volume: 5 µL

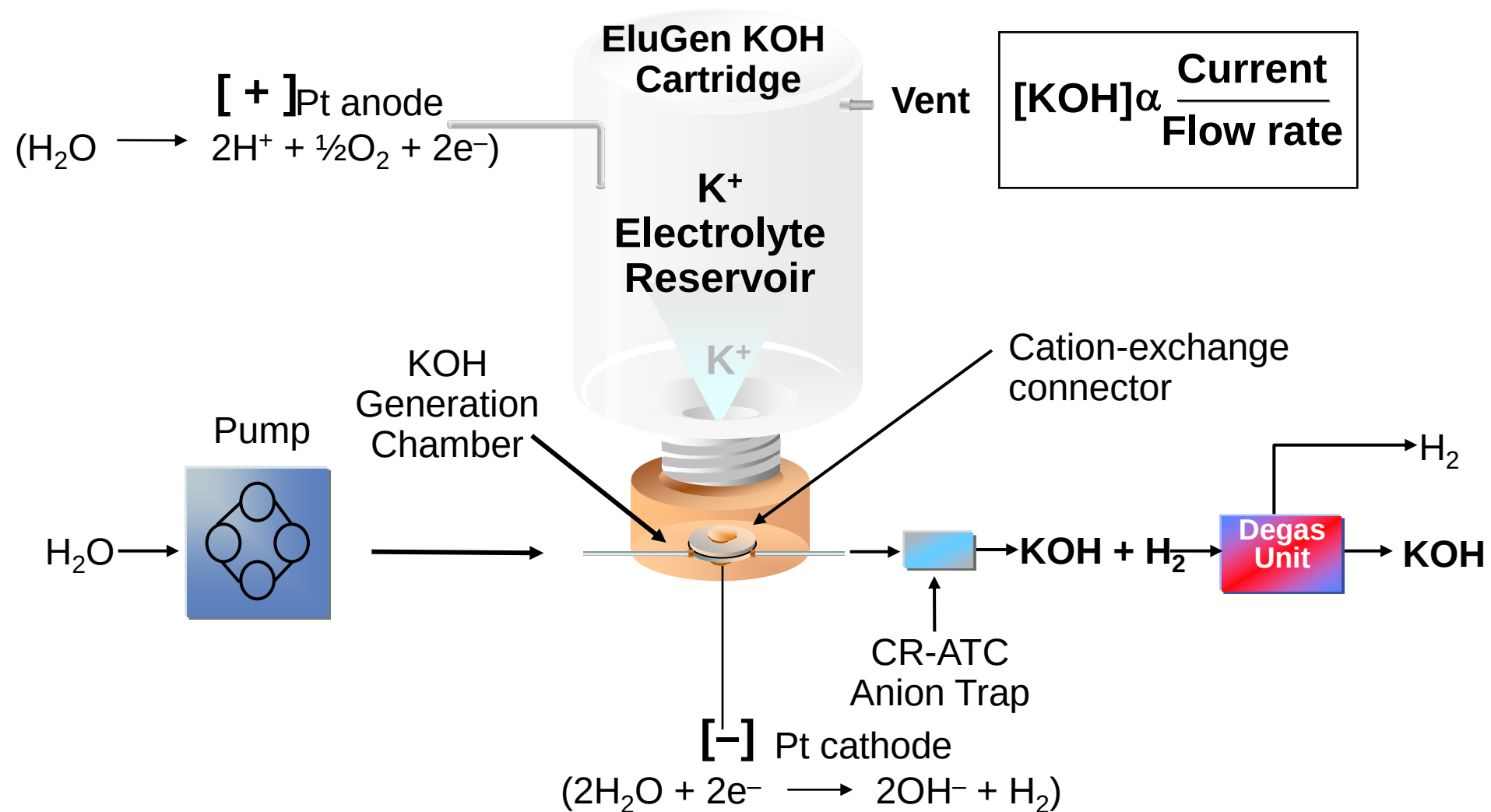
Peaks:

1. Fluoride	2 mg/L
2. Chloride	4
3. Nitrite	10
4. Carbonate	—
5. Bromide	10
6. Sulfate	10
7. Nitrate	10
8. Trifluoroacetate	10
9. Phosphate	20

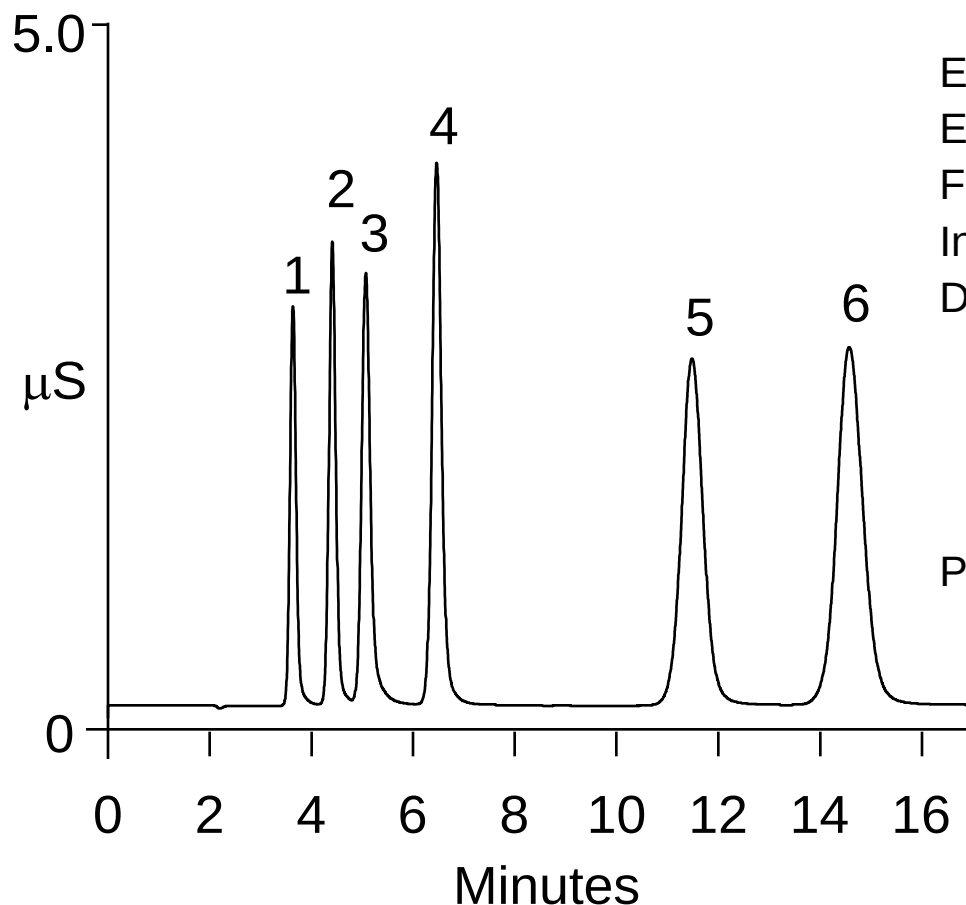
The Role of Chemical Suppression (KOH)



Hydroxide Eluent Generation



Separation of the Common Cations



Column: Dionex IonPac CG12A, CS12A, 4 mm

Eluent: 18 mM Methanesulfonic acid

Eluent Source: Eluent Generator

Flow Rate: 1.0 mL/min

Injection: 25 mL

Detection: Suppressed conductivity, Thermo Scientific™ Dionex™ CSRS™-ULTRA cation self regenerating suppressor, recycle mode

Peaks:

- | | |
|--------------|----------------|
| 1. Lithium | 0.5 mg/L (ppm) |
| 2. Sodium | 2.0 |
| 3. Ammonium | 2.5 |
| 4. Potassium | 5.0 |
| 5. Magnesium | 2.5 |
| 6. Calcium | 5.0 |

- **Conductivity:**
 - **Suppressed**
 - (Non suppressed)
- **UV detection:**
 - **Direct**
 - (Indirect)
 - **Post column derivatization**
- **Amperometry:**
 - Direct current amperometry (DC)
 - **Integrated amperometry (PAD and IPAD)**
- **Mass spectrometry**

- **Anions:** Chloride, sulfate, fluoride, **nitrite**, **nitrate**, bromide, iodide, bromate, chlorite, chlorate, perchlorate, sulfite, thiosulfate, cyanide, thiocyanate, cyanate, sulfide, benzoate, acetate, formate, silicate, glycolate, oxalate, iodate, lactate, trifluoroacetate, numerous other organic acids and inorganic anions, **carbohydrates**, amino acids
- **Cations:** **Lithium**, sodium, potassium, ammonium, **calcium**, magnesium, barium, strontium, methylamine, dimethylamine, trimethylamine, ethanolamine, diethanolamine, triethanolamine, choline, many transition metals, and numerous amines

What IC Offers for Pharmaceutical Analysis

- Easy (direct) determination of analytes lacking chromophores
- Opportunity to have more automated assays compared to HPLC
- Usually requires no organic solvents
- Separation modes better suited for some analytes
- Counterion analysis of salt form drug substances to confirm ID and API content

How IC has been Used in Pharmaceutical Analysis

- Assay
- Determination of impurities and degradation products – limit tests and related substances tests in drug substances and drug products
- Counterion analysis of salt form drug substances to confirm ID and API content
- Excipient analysis

Example IC Methods in the USP-NF

- Assays of Kanamycin B and Amikacin in DS and DP monographs
- USP-NF <345> Assay for Citric Acid/Citrate and Phosphate
- Risedronate Sodium Assay
- Cefepime Hydrochloride—Limit of *N*-methylpyrrolidine
- Methacholine Chloride – Assay and limit of Acetylcholine Test
- Heparin Sodium – Organic Impurities Test
- Sodium Bicarbonate – Limit of Ammonia Test

- Published in 2015
- Eliminated flame tests
- Eliminated wet chemical tests that yielded poor results
- Added better wet chemical tests – EP harmonization too
- Added instrumentation options for identification tests – including ion chromatography and other forms of chromatography.

- The method was published Pharmacopeia Forum 40(5) as part of a modernization proposal for the USP Sodium Nitrite monograph.
- Sodium nitrite part of the treatment for acute cyanide poisoning.
- The IC method assays nitrite and would replace a titration with potassium permanganate.
- The same method determines nitrate impurity.
- We replicated the proposed method in our laboratory, though we used eluent generation.

IC Separation – Sodium Nitrite USP Monograph

Column: Dionex IonPac AS12A Analytical, 4 x 250 mm
Dionex IonPac AG12A Guard, 4 x 50 mm

Eluent: 2.7 mM K₂CO₃ /0.3 mM KHCO₃

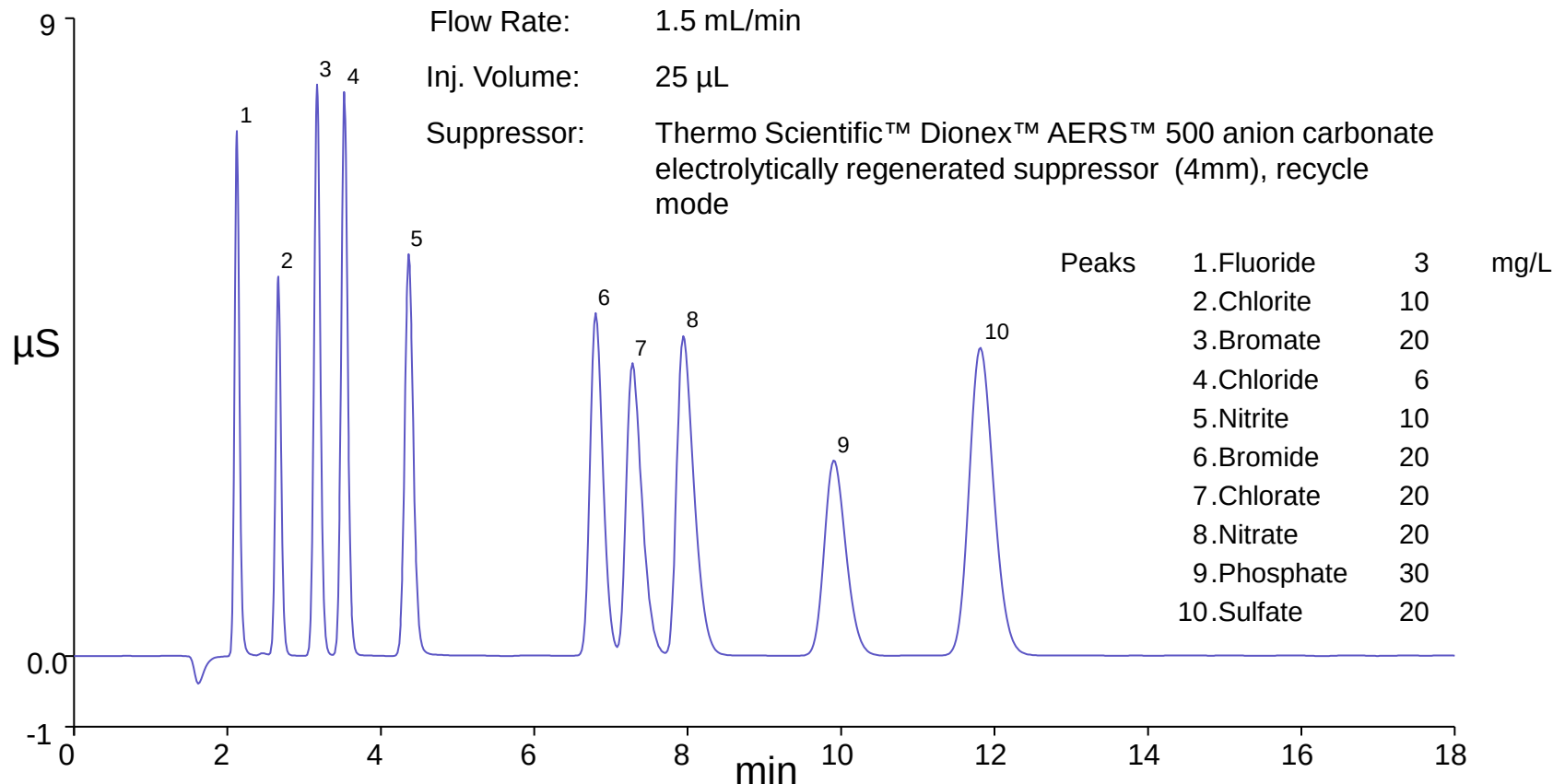
Eluent Source: Thermo Scientific™ Dionex™ Eluent Generator Cartridge
EGC 500 K₂CO₃ cartridge with EPM 500

Temperature: Ambient (~24 °C)

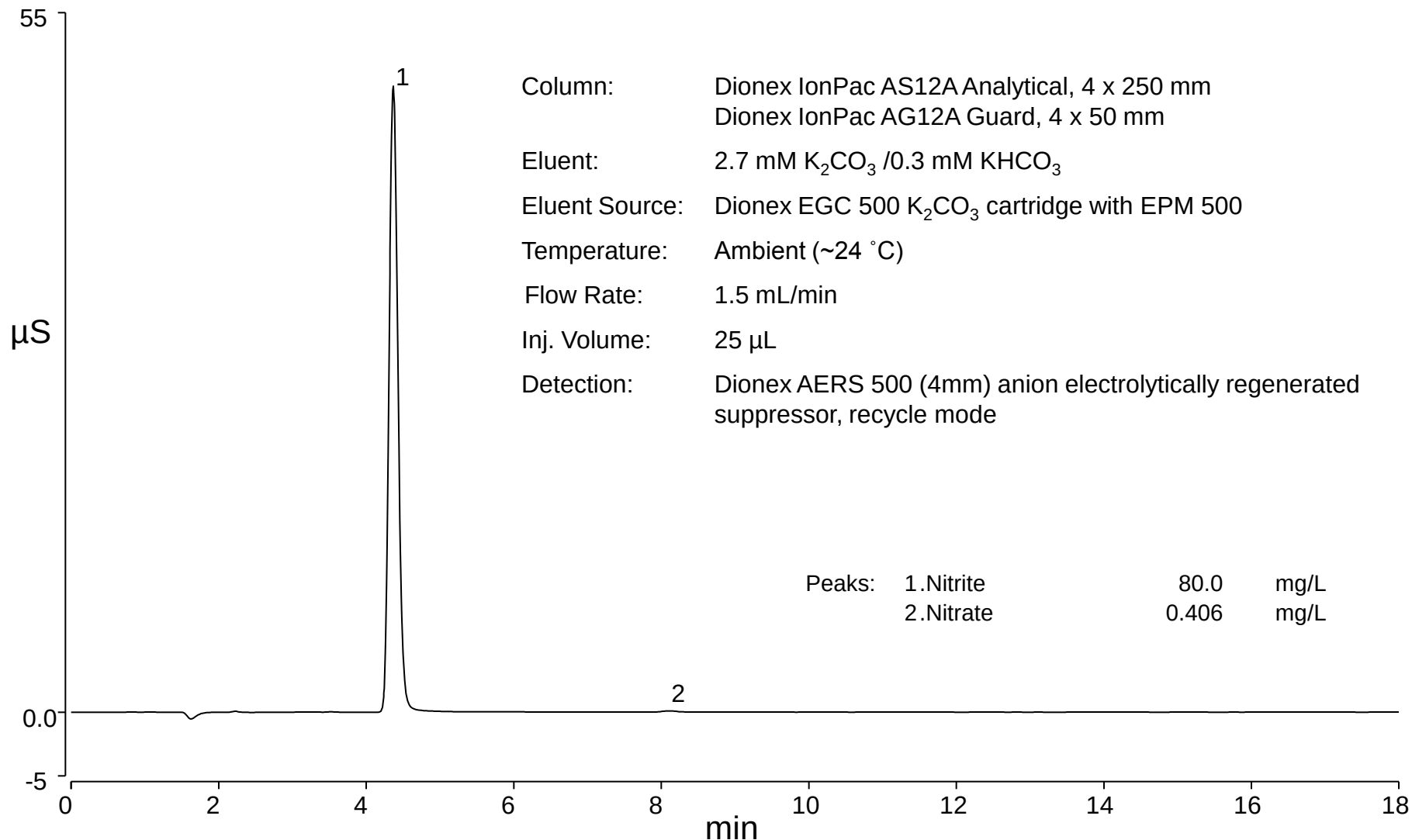
Flow Rate: 1.5 mL/min

Inj. Volume: 25 µL

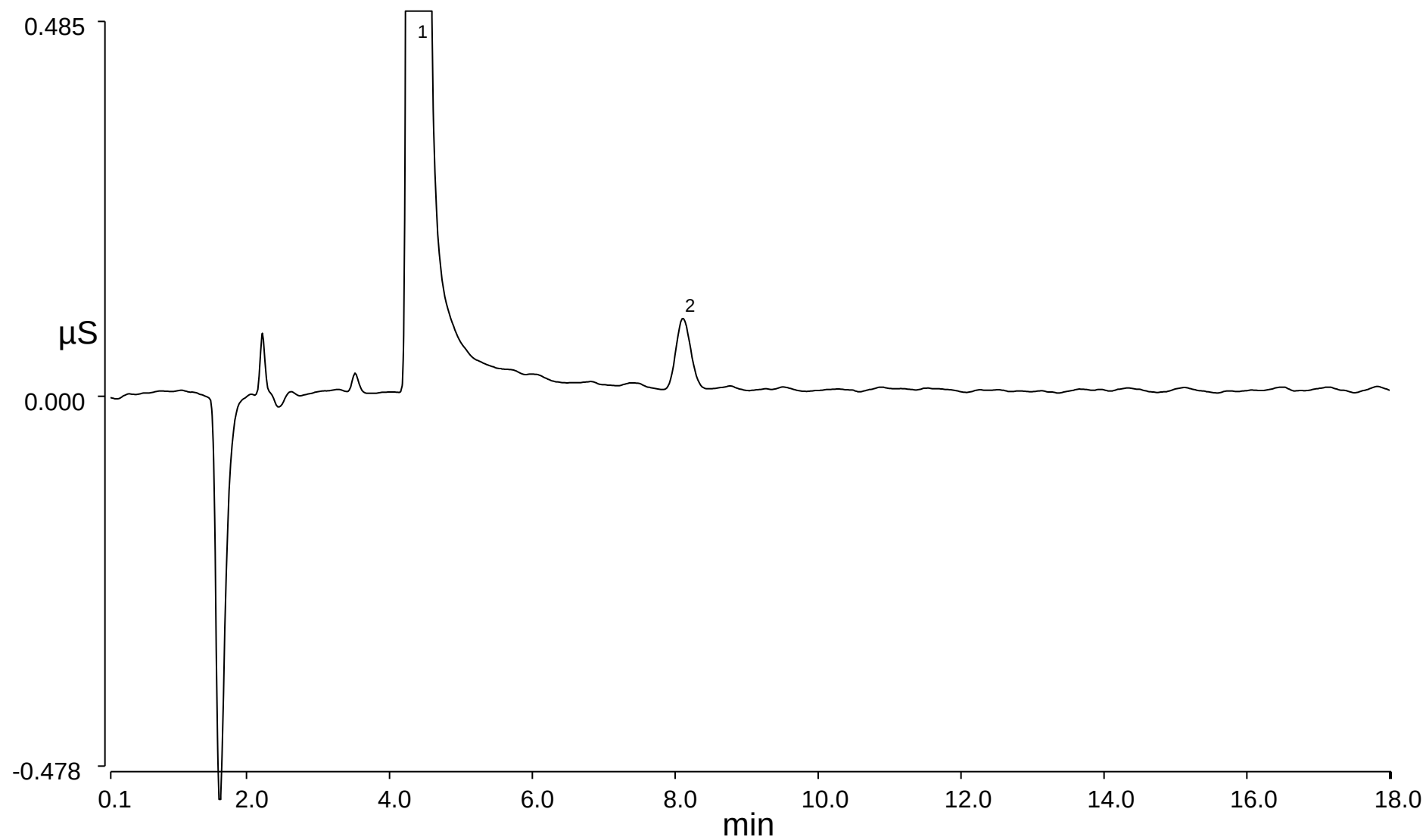
Suppressor: Thermo Scientific™ Dionex™ AERS™ 500 anion carbonate
electrolytically regenerated suppressor (4mm), recycle
mode



Sodium Nitrite Assay by the Proposed USP Monograph



Enlarged to View the Nitrate Peak



Assay of Lithium in Lithium Hydroxide

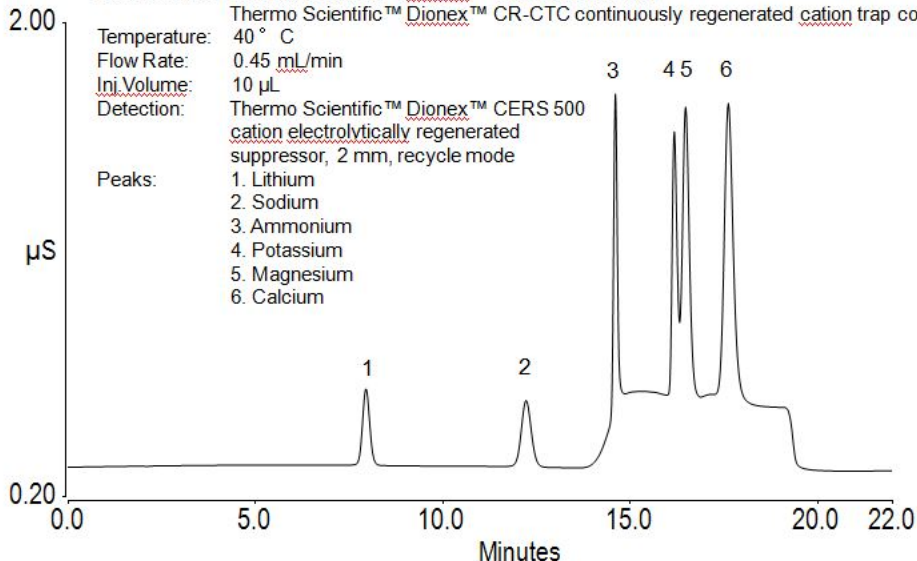
- The method to assay lithium was developed in our lab for a proposal to modernize the USP Lithium Hydroxide monograph.
- The same method was developed to allow the measurement of calcium that is also required in the LiOH monograph.
- Our work has been reported in Application Note 1144.

Separation of Six Common Cations

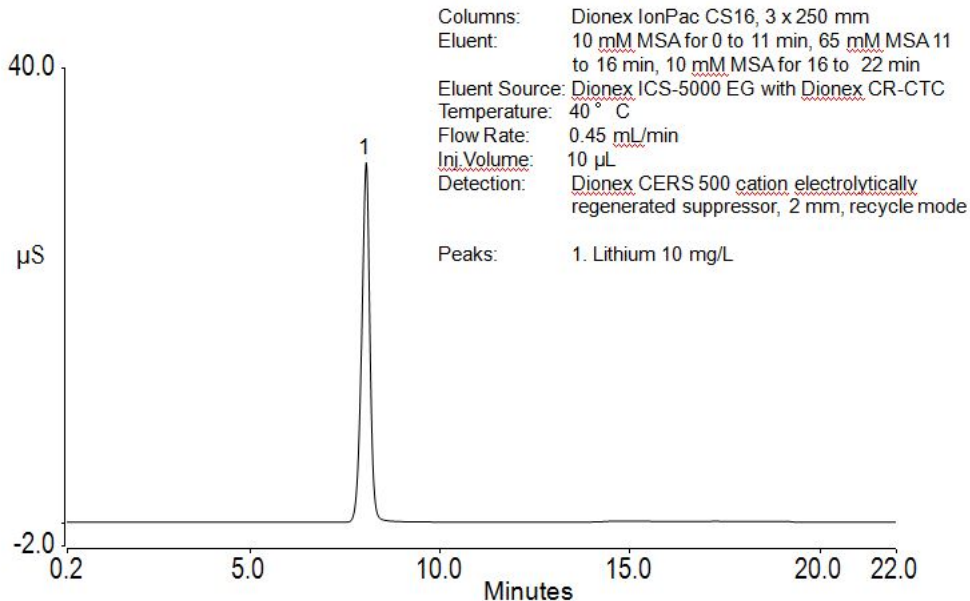
Columns: Dionex IonPac CS16, 3 x 250 mm
Eluent: 10 mM MSA for 0 to 11 min, 65 mM MSA 11 to 16 min, 10 mM MSA for 16 to 22 min
Eluent Source: Thermo Scientific™ Dionex™ ICS-5000 EG with
Thermo Scientific™ Dionex™ CR-CTC continuously regenerated cation trap column
Temperature: 40 ° C
Flow Rate: 0.45 mL/min
Inj. Volume: 10 µL
Detection: Thermo Scientific™ Dionex™ CERS 500
cation electrolytically regenerated
suppressor, 2 mm, recycle mode

Peaks:

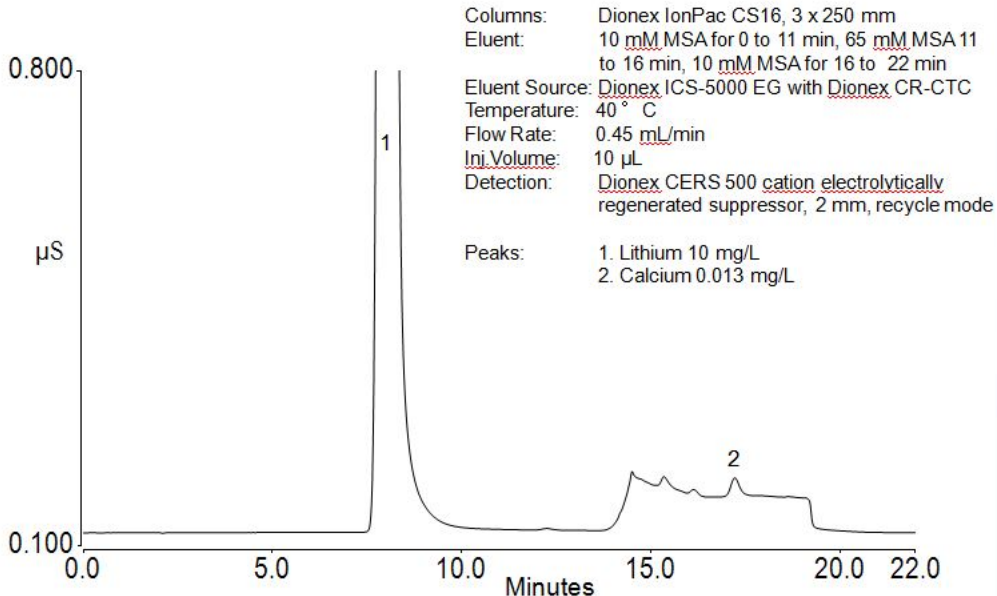
1. Lithium
2. Sodium
3. Ammonium
4. Potassium
5. Magnesium
6. Calcium



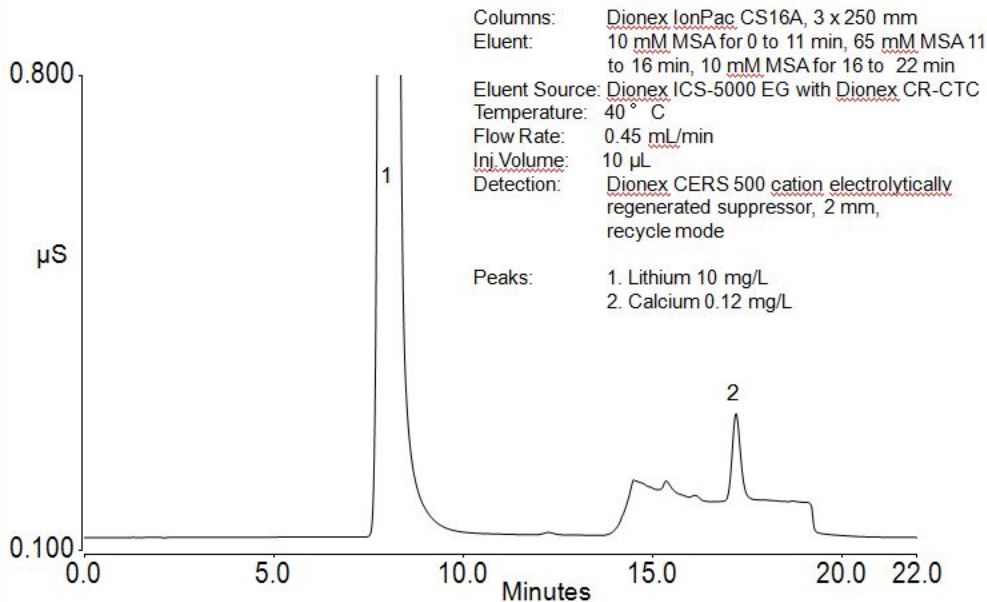
Determination of 10 mg/L Lithium in 10 mM Acetic Acid



Enlarged to View the Calcium Peak



Lithium Sample Spiked with Calcium at the USP Limit

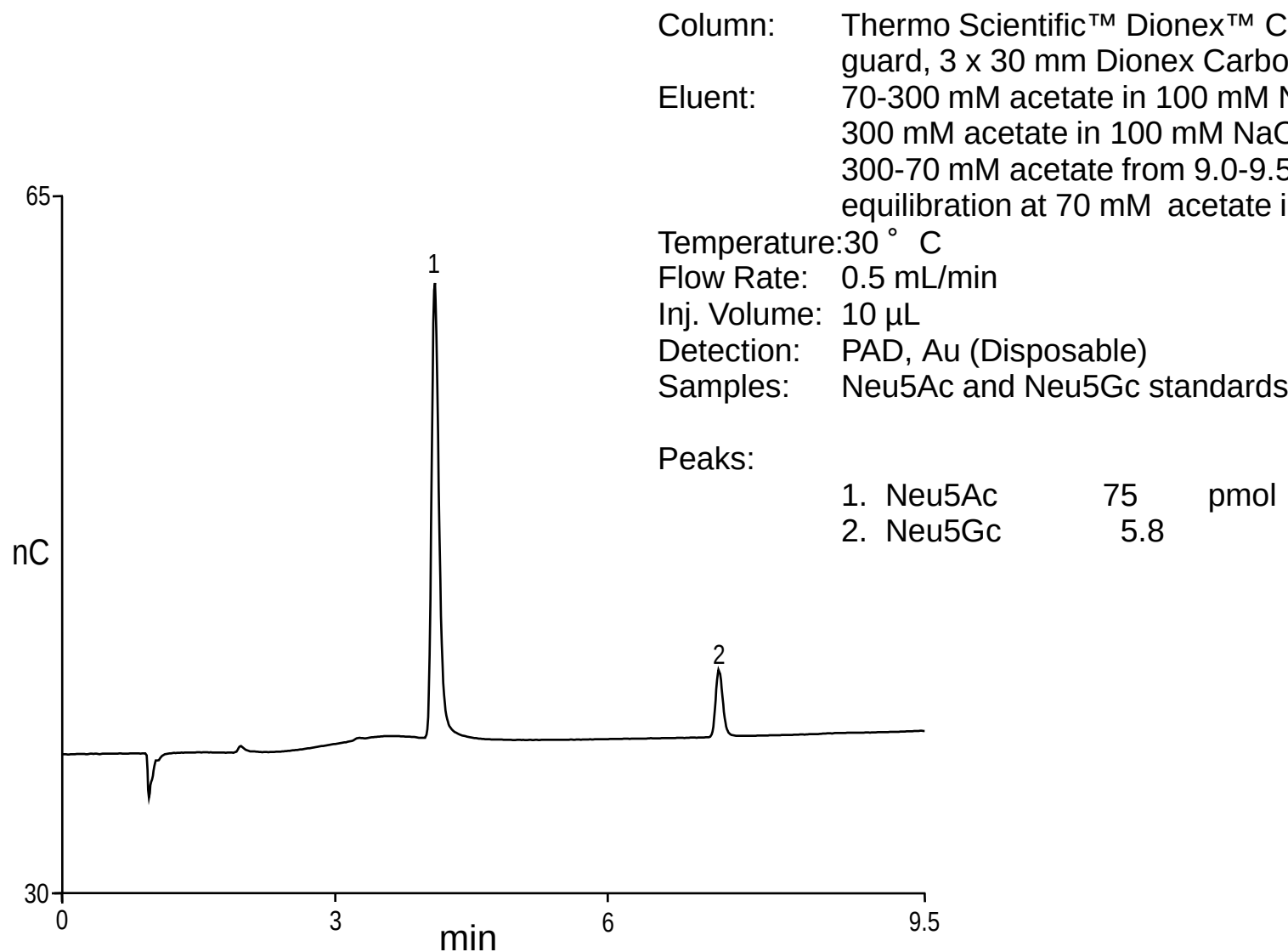


Determination of the Sialic Acid Content of Glycoconjugates

- Common assay for therapeutic glycoproteins.
- One method is an IC (HPAE-PAD) method and it is described in USP General Chapter <129>.

- Sialic acids are released from glycoproteins by either mild acid hydrolysis or by treatment with a neuraminidase.
- Samples are then dried to remove the acid.
- Samples are injected onto the HPAE-PAD system.
- For neuraminidase digestions the sample is either injected or diluted and injected.

Separation of Sialic Acids using HPAE-PAD



Separation of Glycoprotein Acid Hydrolyzates

Column: Dionex CarboPac PA20 guard, 3 x 30 mm
 Dionex CarboPac PA20, 3 x 150 mm

Eluent: 70-300 mM acetate in 100 mM NaOH from 0-7.5 min, 300 mM acetate in 100 mM NaOH from 7.5-9.0 min, 300-70 mM acetate from 9.0-9.5 min. 7 min of equilibration at 70 mM acetate in 100 mM NaOH

Temperature: 30 ° C

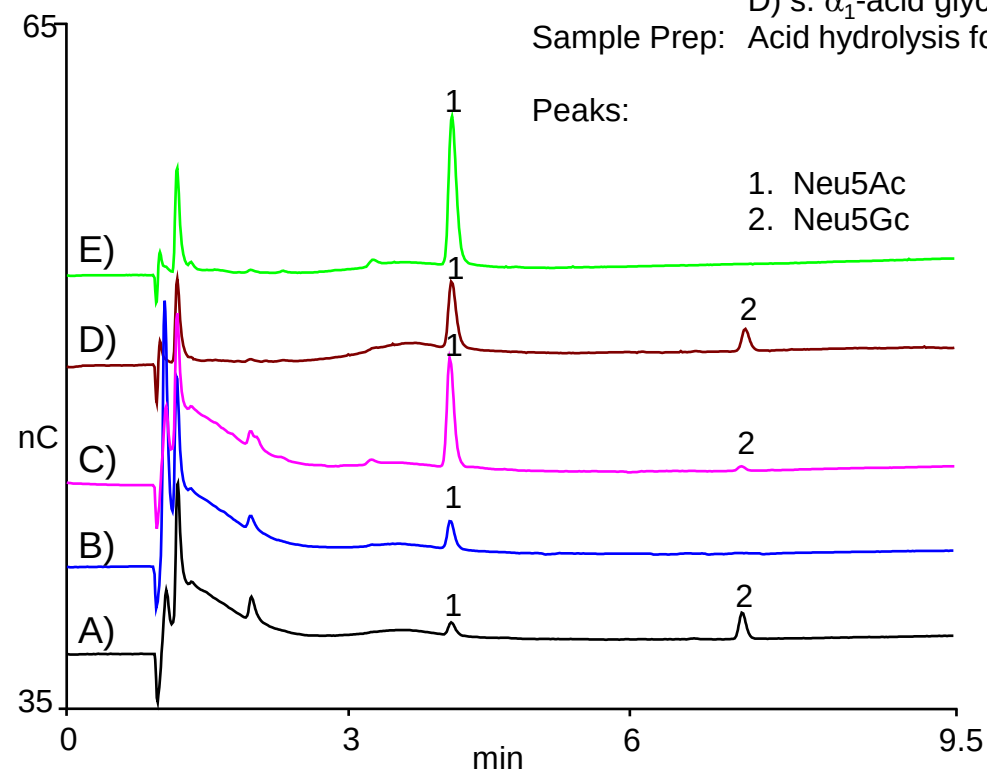
Flow Rate: 0.5 mL/min

Inj. Volume: 10 µL

Detection: PAD, Au (Disposable)

Samples: A) b. apo-transferrin, B) h. transferrin, C) fetuin, D) s. α_1 -acid glycoprotein, E) h. α_1 -acid glycoprotein

Sample Prep: Acid hydrolysis followed by lyophilization and dissolution



Peaks:

1. Neu5Ac
 2. Neu5Gc

A)	B)	C)	D)	E)
1.7	4.4	18	15	37 pmol
2.1	ND	0.39	2.6	ND

A 10% signal offset has been applied.
 ND = Not Detected

- IC is finding greater application in pharmaceutical laboratories to develop methods for drug products and drug substances.
- IC methods have a greater degree of automation compared to other chromatographic techniques.
- IC is one of the techniques being used to modernize pharmacopeia methods.

Acknowledgements

- Jingli Hu
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- Sachin Patil

**Applications of
ION CHROMATOGRAPHY
for PHARMACEUTICAL
and BIOLOGICAL
PRODUCTS**

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Jeffrey S. Rohrer

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Thank you for your attention!